

ORGANIC CONDUCTORS. THE SYNTHESIS OF DI-(2,5-DIHYDROTHIENO)-1,4,5,8-TETRASELENAFULVALENE (DITSF) AND ITS CHARGE-TRANSFER SALT WITH 7,7',8,8'-TETRACYANO-p-QUINODIMETHANE (TCNQ).

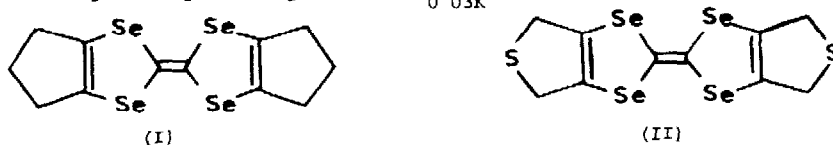
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Several charge-transfer salts of sulphur- and selenium-containing heterofulvalenes with 7,7',8,8'-tetracyano-p-quinodimethane (TCNQ) have recently been shown to possess unique solid state properties<sup>3,4,5,6,7</sup>

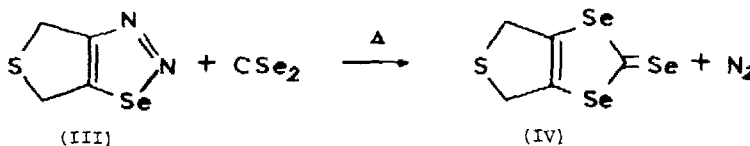
Of the tetraselenafulvalenes prepared so far, HMTSF (I) has been of particular interest because HMTSF-TCNQ<sup>6</sup> is the first example of an organic compound which retains high electrical conductivity at very low temperatures ( $\sigma_{0,03K} \sim 600 (\Omega \text{ cm})^{-1}$ )



Di-(2,5-dihydrothieno)-(3,4-b,3,4-h)-1,4,5,8-tetraselenafulvalene, DITSF (II), was synthesized in order to investigate the effect of introducing a polarisable heteroatom into the exocyclic rings, while retaining the overall size and symmetry, and thereby probably the packing properties, of the congener (HMTSF)

1,2,3-thiadiazoles are known to react with a 1,3-dipolarophile such as  $\text{CS}_2$  to give 1,3-dithiole-2-thiones in low to moderate yields<sup>8</sup>. Recently it was shown that 1,2,3-selenadiazoles can react in an analogous manner to form 1,3-thiaselenole-2-thiones, which in turn were converted to dithiadiselenafulvalenes<sup>9</sup>

We report the reaction of 4,6-dihydro-(3,4-d)-thieno-1,2,3-selenadiazole (III) with  $\text{CSe}_2$ <sup>10</sup> to yield the 1,3-diselenole-2-selone (IV)



(IV) was treated with triethylphosphite to give DTTSF (II).

(III) was prepared from dihydrothiophene-3-one semicarbazone<sup>11</sup> by ring closure with  $\text{SeO}_2$  in glacial acetic acid by a procedure analogous to that of Yalpani *et al*<sup>12</sup>. (III) was isolated as colourless needles in 18% yield, m p  $85-87^\circ$  (hexane), and identified as being (III) rather than the other possible isomer, 4,5-dihydro-(2,3-d)-thieno-1,2,3-selenadiazole, by  $^1\text{H-NMR}$  spectroscopy.

Reaction of (III) with excess  $\text{CSe}_2$  in boiling toluene (2h) afforded (IV) in 35% yield, m p  $165-166^\circ$  MS  $\text{M}^+$  in accordance with the calculated isotopic cluster for  $\text{C}_5\text{H}_4\text{SSe}_3$  NMR( $\text{CDCl}_3$ ) 4.07(s)  $\delta$  relative to TMS

Triethylphosphite coupling of (IV) in toluene (24h,  $100^\circ$ ) gave the insoluble (II) in 7-10% yield as black crystals, m p  $228-230^\circ$  (decomp) MS  $\text{M}^+$  in accordance with the calculated isotopic cluster for  $\text{C}_{10}\text{H}_8\text{S}_2\text{Se}_4$  UV( $\text{C}_6\text{H}_5\text{CN}$ ),  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 317 (3.92), 438 (2.70) and 495 (2.72) nm

(II) forms a highly conducting 1:1 charge-transfer salt with TCNQ. From chlorobenzene black needles of typical dimensions  $1 \times 0.02 \times 0.004$  mm were obtained. Room-temperature conductivities of at least  $600 (\Omega \text{ cm})^{-1}$  were measured by 4-probe techniques. A detailed account of the temperature-dependent electrical properties will appear elsewhere.

#### References and Notes

1. Kemisafdelingen, Risø, DK-4000, Denmark
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3. For a review, see A. F. Garito and A. J. Heeger, *Acc Chem Res*, **7**, 232 (1974)
4. E. M. Engler and V. V. Patel, *J Amer Chem Soc*, **96**, 7376 (1974)
5. K. Bechgaard, D. O. Cowan and A. N. Bloch, *J Chem Soc, Chem Commun*, 937 (1974)
6. A. N. Bloch, D. O. Cowan, K. Bechgaard, R. E. Pyle, R. H. Banks and T. O. Poehler, *Phys Rev Letters*, **34**, 1561 (1975)
7. (a) M. V. Laksmikantham, M. P. Cava and A. F. Garito, *J Chem Soc, Chem Commun*, 383 (1975)  
(b) E. M. Engler and V. V. Patel, *ibid*, 671 (1975)
8. R. Huisgen and V. Weberndörfer, *Experientia*, **17**, 566 (1961)
9. H. K. Spencer, M. V. Laksmikantham, M. P. Cava and A. F. Garito, *J Chem Soc, Chem Commun*, 867 (1975)
10. L. Henriksen and E. S. Christiansen, *Int J Sulfur Chem Part A*, **2**, 13 (1972)
11. P. Karrer and H. Schmid, *Helv Chim Acta*, **27**, 116 (1944)
12. I. Lalezari, A. Shafiee and M. Yalpani, *J Org Chem*, **36**, 2836 (1971)